

## MATERIAL SAFETY DATA SHEET

Section I – Product Information						
Product Name or Identity:	Middlebrook 7H11 Agar					
Manufacturer's Name:	Acumedia Manufacturers, Inc. Emergency Phone No. 517/372-9200					
	740 East Shiawassee         Fax No.:         517/372-2006					
Lansing, Michigan 48912 e-mail: foodsafety@neogen.com						
Date Prepared or Revised: 10/10/04						

Section II – Hazardous Ingredients / Identity Information					
Hazardous Components: (Specific Chemical Identity: Common Names)	OSHA PEL (Permissible Exposure Limits)	ACGIH TLV (Threshold Limit Value)	Toxicity Data LD₅₀		
Sodium Citrate	N/A	N/A	N/A		
Ammonium Sulfate	N/A	N/A	ORL-RAT, 2840 mg/kg		
Sodium Phosphate	N/A	N/A	ORL-RAT, 17000 mg/kg		
Potassium Phosphate	N/A	N/A	SKN-RBT, > 4640 mg/kg		

Section III – Physical Characteristics				
Boiling Point: N/A         Specific Gravity (H <sub>2</sub> O = 1): 1.7 (Sodium Citrate), 1.67 (Sodium Phosphere)				
Vapor Pressure (mm Hg.): N/A         Melting Point: 150°C (Sodium Citrate), > 465°C (Potassium Phosphate)				
	240°C (Sodium Phosphate), 235 - 280°C (Ammonium Sulfate)			
Vapor Density (AIR = 1): N/A	Evaporation Rate (Butyl Acetate = 1): N/A			
Solubility in Water: 72 g/100 g of water (Sodium Citrate), 150 g/ 100 g cold water (Potassium Phosphate)				
Soluble (Sodium Phosphate), 41.22 g /100 g water (Ammonium Sulfate)				
Appearance and Odor: White crystals, odorless (Sodium Citrate), White crystals or powder, odorless (Potassium Phosphate)				

Appearance and Odor: White crystals, odorless (Sodium Citrate), White crystals or powder, odorless (Potassium Phosphate) White powder (Sodium Phosphate), White granules or crystals, slight odor of ammonia (Ammonium Sulfate).

Section IV – Fire and Explosion Hazard Data				
Flash Point (Method Used): Not applicable       Flammable Limits: LEL (Lower Explosive Limit) - N/A				
	UEL (Upper Explosive Limit) - N/A			
Extinguishing Media: Suitable extinguishing agents. CO <sub>2</sub> , extinguishing powder, or water spray				

**Special Fire Fighting Procedures:** Fight larger fires with water or alcohol resistant foam. Firefighters should wear protective equipment and self-contained breathing apparatus. As with most organic solids, fire is possible at elevated temperatures or by contact with an ignition source.

**Unusual Fire and Explosion Hazards:** During heating or in case of fire, poisonous gases are produced. Fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust explosion hazard.

	Section V – Reactivity Data												
Stability:	Unstable				Avoid:	Stabl	e under	ordinary	conditions	of	use	and	storage.
			Hygros	copic.									
	Stable	Х											
	Incompatibility (Materials to Avoid): Reacts with oxidizing agents, Potassium chlorate, Potassium nitrite, Sodium-												
Potassium powder, and Sodium hypochlorite.													
Hazardous Decomposition or Byproducts: Carbon monoxide (C0), Carbon dioxide (CO <sub>2</sub> ), and Sodium oxide may form						may form							
when heated to decomposition. May emit ammonia.													
Hazardous	May Occ	ur		Condi	tions to	Avoid:	Heat, fla	ame, igniti	on sources,	dus	st and	d inco	ompatible
Polymerizatio	on:			mater	ials.			-					
	Will Not 0	Occur	Х										

Section VI – Health Hazard Data							
Route(s) of Entr	y: Inhalation? Yes	Skin? Yes	Ingestion? Yes				
Health Hazards: (Acute and Chroi							
Carcinogenicity	Carcinogenicity:         NTP? No (National Toxicology Program)         IARC Monographs? No         OSHA Regulated? No						
<ul> <li>Signs and Symptoms of Exposure: If large doses are ingested, can cause vomiting, diarrhea, and prostration Inhalation of large amounts of dust may cause irritation to the respiratory tract. Symptoms may include coughing o shortness of breath. Possible irritation on prolonged contact with moist or sensitive areas of the skin.</li> <li>Medical Conditions Generally Aggravated by Exposure: May be irritating to mucous membranes and respiratory tract. The toxicity of phosphates is their ability to sequester calcium. Chronic exposure of phosphates may sequester calcium and cause calcium phosphates deposits in the kidneys.</li> </ul>							
Emergency / First Aid Procedures:	Ingestion:       If swallowed, seek medical attention.         Inhalation:       Supply fresh air or oxygen. Seek medical attention.						
	Eye Contact: Rinse opened eye for at least 15 minutes under running water. Seek medical attention.Skin Contact: Wash with plenty of soap and water for 15 minutes. Seek medical attention.						

## Section VII – Precautions for Safe Handling and Use

Accidental Release Measures: Ventilate spill area. Avoid raising dust. Wear suitable protective clothing. Wipe up with damp sponge or mop. Avoid contact with skin, eyes, and clothing.

Waste Disposal Method: Dispose in accordance with all applicable federal, state, and local environmental regulations.

**Handling and Storing:** Keep container tightly closed. Protect from moisture and physical damage. Suitable for any general chemical storage area. Store away from oxidizing agents. Containers of this material may be hazardous when empty since they retain product residues.

**Other Precautions:** Prevent formation of dust. Ensure good ventilation / exhaustion at the workplace. Avoid prolonged or repeated exposure.

## Section VIII – Control Measures

**Respiratory Protection** (Specify Type): None required where adequate ventilation conditions exist. If airborne concentration is high, use an appropriate respirator or dust mask.

Ventilation:	Local Exhaust: 50 – 100 CFM	Special: N/A			
	Mechanical (General): N/A	Other: N/A			
Protective Glove	es: Proper disposable gloves	Eye Protection: Chemical resistant safety goggles			

Other Protective Clothing or Equipment: Uniform, lab coat, or disposable lab wear.

**Work / Hygienic Practices:** Follow the usual precautionary measures for handling chemicals / powder. Keep away from food and beverages. Immediately remove all soiled and contaminated clothing. Avoid contact with eyes, skin, and clothing.

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