

MATERIAL SAFETY DATA SHEET

Section I – Product Information				
Product Name or Identity:	Pseudomonas Isolation Agar			
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Manufacturer's Name:	Acumedia Manufacturers, Inc.	Emergency Phone No.	517/372-9200	
	740 East Shiawassee	Fax No.:	517/372-2006	
	Lansing, Michigan 48912	e-mail:	foodsafety@neogen.com	
Date Prepared or Revised: 10/10/04				

Section II – Hazardous Ingredients / Identity Information				
Hazardous Components:	OSHA PEL	ACGIH TLV	Toxicity Data LD ₅₀	
(Specific Chemical Identity: Common Names)	(Permissible	(Threshold	•	
,	Exposure Limits)	Limit Value)		
Magnesium Chloride	N/A	N/A	ORL-RAT, 2800 mg/kg	
Potassium Sulfate	N/A	N/A	ORL-RAT, 6600 mg/kg	

Section III – Physical Characteristics				
Boiling Point: 1670°C (Potassium Sulfate)	Specific Gravity (H ₂ O = 1): 2.32 g/cm ³ (Magnesium Chloride) 2.66 g/cm ³ (Potassium Sulfate)			
Vapor Pressure (mm Hg.): N/A	Melting Point: 714°C (Magnesium Chloride) 1067°C (Potassium Sulfate)			
Vapor Density (AIR = 1): N/A	Evaporation Rate (Butyl Acetate = 1): N/A			
Solubility in Water: 167 g/ 100 mL water at 20°C (Magnesium Chloride), Soluble (Potassium Sulfate)				
Appearance and Odor: Colorless crystals or flakes, odorless (Magnesium Chloride).				
White powder (Potassium Sulfate).				

Section IV – Fire and Explosion Hazard Data				
Flash Point (Method Used): Not applicable	Flammable Limits: LEL (Lower Explosive Limit) - N/A			
	UEL (Upper Explosive Limit) - N/A			
Extinguishing Media: Suitable extinguishing agents. CO ₂ , extinguishing powder, or water spray				
Special Fire Fighting Procedures: Fight larger fires with water or alcohol resistant foam. Firefighters should wear protective equipment and self-contained breathing apparatus.				

Section V – Reactivity Data						
Stability:	Unstable			ons to Avoid: Stable under ordinary conditions of use and storage. By strong , Magnesium Chloride is converted into oxychloride. Hygroscopic.		
	Stable	X				
	Incompatibility (Materials to Avoid): Strong oxidizing agents. When Magnesium Chloride is mixed with limited amount of water, enough heat may be generated to cause frothing.					
Hazardous Decomposition or Byproducts: When heated to decomposition, Magnesium Chloride emits corrosive						
hydrochloric acid vapor. When heated to temperatures above 300°C it emits toxic fumes of chlorine gas.						
Hazardous	May Oc	cur		Conditions to Avoid: Heat, moisture, and incompatible materials.		
Polymerizat	ion:			·		
-	Will Not	Occur	Х			

Section VI – Health Hazard Data						
Route(s) of Entr	ry:	Inhalation? Yes	Skin? Yes	Ingestion? Yes		
Health Hazards: Irritant. Irritating to eyes, respiratory system, and skin. May be harmful if swallowed. (Acute and Chronic)						
Carcinogenicity: NTP? No (National Toxicology Program)			IARC Monographs? No (International Agency for Research in Cancer)	OSHA Regulated? No		
respiratory tract.	Signs and Symptoms of Exposure: Inhalation of dust may cause irritation to mucous membranes and upper respiratory tract. May be harmful if absorbed through the skin, irritation may develop.					
Medical Conditions Generally Aggravated by Exposure: Magnesium salts are slowly absorbed, therefore abdominal pain, vomiting and diarrhea may be the only symptoms. If elimination is blocked by bowel blockage or other reasons, CNS depression, lack of reflexes, or hypocalcemia may occur.						
Emergency / First Aid	Note to Physician: IV administration of calcium gluconate will partially reverse the effects of acute magnesium toxicity. Ventricular support with calcium chloride infusion and mannitol forced diuresis has also been successful.					
Procedures:	Ingestion: If swallowed, seek medical attention immediately.					
Inhalation: Supply fresh air or oxygen. Seek medical attention.						
	Eye Contact: Rinse opened eye for at least 15 minutes under running water. Seek medical attention.					
	Skin Contact: Wash with plenty of soap and water for 15 minutes. Seek medical attention if irritation develops.					

Section VII - Precautions for Safe Handling and Use

Accidental Release Measures: Ventilate spill area. Wear suitable protective clothing. Wipe up with damp sponge or mop. Avoid inhalation, contact with eyes, skin, and clothing.

Waste Disposal Method: Dispose in accordance with all applicable federal, state, and local environmental regulations.

Handling and Storing: Keep container tightly closed. Protect from moisture and physical damage. Store away from incompatible materials. Containers of this material may be hazardous when empty since they retain product residues.

Other Precautions: Prevent formation of dust. Ensure good ventilation / exhaustion at the workplace. Avoid prolonged or repeated exposure.

Section VIII – Control Measures					
Respiratory Protection (Specify Type): None required where adequate ventilation conditions exist. If airly concentration is high, use an appropriate respirator or dust mask.					
Ventilation:	Local Exhaust: 50 – 100 CFM	Special: N/A			
	Mechanical (General): N/A	Other: N/A			
Protective Gloves: Proper disposable gloves		Eye Protection: Chemical resistant safety goggles			
Other Protective Clothing or Equipment: Uniform, lab coat, or disposable lab wear.					

Work / Hygienic Practices: Follow the usual precautionary measures for handling chemicals / powder. Keep away from food and beverages. Immediately remove all soiled and contaminated clothing. Avoid contact with eyes, skin, and clothina.

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